12020 Olivine Basalt 312 grams



Figure 1: Photo of 12020. Scale in cm. NASA # S70-43639.

Introduction

12020 is an olivine basalt with large rounded olivine phenocrysts and elongate pyroxene crystals set in a variolitic groundmass of thin clinopyroxene and plagioclase laths (figure 2). It had one large zap pit on one surface.

Petrography

Klein et al. (1971) describe 12020 as a "mediumgrained olivine microgabbro consisting mainly of clinopyroxene, plagioclase and olivine. The clinopyroxene occurs as subhedral laths, up to several mm in length, as smaller anhderal grains, and as very thin, lathlike, crystals interleathed with plagioclase laths."



Figure 2: Photomicrograph of thin section 12020,13 illustrating round olivine and elongate pyroxene phenocrysts with variolitic intergrowths of plagioclase and pyroxene needles. NASA S70-30254. Field of view about 2 cm.



Figure 3: Photomicrographs of thin section 12020,11 (plane-polarized, crossed-nicols). Field of view is 2.6 mm. NASA # S70-49556-557.

Mineralogy

Olivine: According to Klein et al. (1971) the cores of olivine in 12020 are rather homogeneous (Fo_{70-77}), whereas the rims range from Fo_{70} to Fo_{50} (figure 4). Kushiro et al. (1971) reported a wide range in olivine composition from Fo_{74} to Fo_3 .

Pyroxene: Kushiro et al. (1971) and Klein et al. (1971) studied the composition of pyroxene in 12020 (figure 4). Pyroxene zones in Fe all the way to ferroheddenbergite (now there's a name).

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Figure 4: Pyroxene and olivine composition of 12020 (adapted from Klein et al. 1971, Kushiro et al. 1971).

Plagioclase: Plagioclase composition in 12020 range from An_{93} to An_{88} (Kushiro et al. 1971). Klein et al. (1971) report An_{98} to An_{80} , with the majority as An_{96} .

Chemistry

The rare earth element pattern is relatively flat (figure 5). 12020 is relatively Mg-rich (figure 6), apparently due to accumulation of olivine (Walker et al. 1976).

Radiogenic age dating

Alexander et al. (1972) reported an Ar/Ar age of 3.20 \pm 0.03 b.y. for 12020, consistent with that of other Apollo 12 basalts.

Cosmogenic isotopes and exposure ages

Hintenberger et al. (1971) determined exposure ages for 12020 using ³He (77 m.y.), ²¹Ne (71 m.y.) and ³⁸Ar (56 m.y.). The suntan age for 12020 (from etched solar flare track studies) is 2.6 m.y. (Bhandari et al. 1971).

Other Studies

Bogard et al. (1971) reported the content and isotopic composition of rare gases in 12020.

There are 13 thin sections.

Willieralogical Wode for 12020							
-	Neal et	Klein et	Papike et				
	al. 1994	al. 1971	al. 1976				
Olivine	19	15.1	11.4				
Pyroxene	51.2	58.6	61.4				
Plagioclase	25.9	20	20.7				
Opaques			5.6				
Ilmenite	0.2						
Chromite +Usp	2.7	4.6					
Mesostasis	0.5	1.7					
"silica"			0.2				

		. 1000	
List of Photo #	#s for 12020		12020
S69-24225 S69-24213 S69-64130 S69-64105	TS closeup color mug	100 sample/ chondrite	
S69-63324-332 S70-43638-640 S70-19641-644	B&W mug color mug wire saw cut	10	*
S70-49135-144 S70-25406-408 S70-25421-424	TS color	1	
S70-25890-893 S70-25890-893 S70-27991		0.1	La Ce Pr Nd Sm Eu ^{Gd} Tb ^{Dy} Ho ^{Er} Tm ^{Yb} Lu
S70-31559-566	TS		Figure 5: Para earth alament diagram for 120





Figure 6: Composition of lunar basalts showin, 12020.

Table 1. Chemical composition of 12020.

reference weight	Kushiro	o71	Hubbar	d71	Weisman	n75	Wanke	71	Cuttitta	171	Wakita	171		Haskin7	71	Compsto	n71	Anders	71
SiO2 % TiO2 Al2O3 FeO MnO MgO CaO Na2O K2O P2O5 S % sum	44.45 2.54 7.99 20.65 0.26 14.89 8.53 0.21 0.06 0.02	(a) (a) (a) (a) (a) (a) (a) (a) (a)	0.19 0.056	(b)	0.056	(b)	43.86 2.64 7.2 21.1 0.28 15.65 8.12 0.17 0.046	(c) (c) (c) (c) (c) (c) (c) (c)	44.6 2.56 8 20.7 0.27 14.4 8.53 0.23 0.06 0.08	(d) (d) (d) (d) (d) (d) (d) (d) (d) (d)	42.2 2.7 8.5 21.8 0.253 16.1 8.7 0.22	3 8.3 0.26 8.8 0.213 0.069	(c) (c) (c) (c) (c) (c) (c) (c)			44.66 2.73 7.31 21.58 0.28 13.91 8.73 0.21 0.064 0.08 0.06	(f) (f) (f) (f) (f) (f) (f) (f) (f) (f)		
Sc ppm V							45.4	(c)	39 155	(d) (d)	42 180	200	(c) (e)			146	(f)		
Cr Co Ni	4653	(a)					4560 61	(c) (c)	4330 64 77	(d) (d) (d)	4187 61		(c) (c)			3780 50 50	(f) (f) (f)	68	(e)
Cu Zn							6.9	(c)	9	(d)						13 4	(f) (f)	0.74	(e)
Ga Ge ppb									4.8	(d)						1.8	(f)		
Se Rb Sr Y Zr Nb Mo Ru			0.997 93.6	(b) (b)	0.997 93.6	(b) (b)			1.4 65 37 119 13	(d) (d) (d) (d) (d)		1 34	(e) (e)			1.03 91.4 32 97 5	(f) (f) (f) (f)	0.114 0.85	(e) (e)
Rh Pd ppb Ag ppb Cd ppb In ppb Sn ppb Sb ppb To ppb												12	(e)					0.98 1.1 2.7	(e) (e) (e)
Cs ppm Ba			64.4	(b)	64.4	(b)			61	(d)	25	0.05	(e) (c)			60	(f)	0.039	(e)
La Ce			16.1	(b)	16.1	(b)	4.82	4.82 (c) 3.4 (c) 0.82 (c) 0.91 (c) 5.68 (c) 0.7 (c)			5.9 4.08 0.79	5.6 16.1 2.2 12 4.1 0.76 5.4		(c) (c)	:)4 :)11	(f) (f)			
Pr Nd Sm Eu Gd			12 4.5 0.839 5.43	(b) (b) (b) (b)	12 4.5 0.839 5.43	(b) (b) (b) (b)	3.4 0.82))					13 3.92 0.76 5.3	13 (c) 3.92 (c) 0.76 (c) 5.3 (c))))			
Tb Dy Ho			6.13	(b)	6.13	(b)	0.91 5.68 1.07		(c) (c)				0.96 5.2	(e) (e)	1.02 7.1 1.34	(c) (c)			
Er Tm			3.75	(b)	3.75	(b)	1.07	(0)			3.5 0.59	(e) (e) (e)	3.8	(c) (c)					
Yb Lu Hf Ta W ppb Re ppb			3.69	(b)	3.28 0.14	(b) (b)	2.91 0.42 3.8 0.45	(c) (c) (c) (c)	5.1	(d)	3.8 0.54 2.4	3.7 0.52	(e) (e) (c)	3.43 0.51	(c) (c)				
Os ppb Ir ppb Pt ppb																		0.04	(e)
Au ppb U ppm <i>technique</i>	(a) con	venti	ional we	t, (b)) IDMS, (c) IN	'AA , (d)	mixe	ed mcirc	cher	n, XRF,	emissior	(c) n spe	ec., (e) R	NAA	A, (f) XRF		0.36	(e)

reference weight SiO2 %	Neal2001		gas analysis on neutron irradiated Apollo 12 samples. <i>Proc.</i> <i>3rd Lunar Sci. Conf.</i> 1787-1795.
TiO2 Al2O3 FeO MnO MgO CaO Na2O K2O P2O5			Anders E., Ganapathy R., Keays R.R., Laul J.C., and Morgan J.W. (1971) Volatile and siderophile elements in lunar rocks: Comparsion with terrestrial and meteoritic basalts. <i>Proc.</i> 2 nd Lunar Sci. Conf. 1021-1036.
S % sum	47	(a)	Anders E., Ganapathy R., Krahenbuhl U. and Morgan J.W. (1973) Meteoritic material on the Moon. <i>The Moon</i> 8 , 3-24.
V Cr Co Ni Cu Zn Ga Ge ppb As Se	182 3615 66 89 16 89 3.35	 (a) (a) (a) (a) (a) (a) 	Bhandari N., Bhat S., Lal D., Rajagopalan G., Tamhane A.S. and Venkatavaradan V.S. (1971) High resolution time averaged (millions of years) energy spectrum and chemical composition of iron-group cosmic-ray nuclei at 1 A.U. based on fossil tracks in Apollo samples. <i>Proc. 2nd Lunar Sci.</i> <i>Conf.</i> 2611-2619.
Sr Y Zr Nb Mo Ru Rh	1.13 105 43 119 8.45 1.12	(a) (a) (a) (a) (a)	Bogard D.D., Funkhouser J.G., Schaeffer O.A. and Zahringer J. (1971) Noble gas abundances in lunar material-cosmic ray spallation products and radiation ages from the Sea of Tranquillity and the Ocean of Storms. <i>J. Geophys.</i>
Pd ppb Ag ppb Cd ppb In ppb Sn ppb Sb ppb Te ppb	30	(a)	Compston W., Berry H., Vernon M.J., Chappell B.W. and Kay M.J. (1971) Rubidium-strontium chronology and chemistry of lunar material from the Ocean of Storms. <i>Proc.</i> 2 nd Lunar Sci. Conf. 1471-1485.
Cs ppm Ba La Ce Pr Nd Sm	0.02 64 5.87 17.6 2.53 12.9 4.38	 (a) (a) (a) (a) (a) (a) (a) 	Haskin L.A., Helmke P.A., Allen R.O., Anderson M.R., Korotev R.L. and Zweifel K.A. (1971) Rare-earth elements in Apollo 12 lunar materials. <i>Proc.</i> 2 nd Lunar Sci. Conf. 1307-1317.
Eu Gd Tb Dy Ho Er	0.73 5.78 1.11 7.11 1.41 3.82	(a) (a) (a) (a) (a)	Hintenberger H., Weber H.W. and Takaoka N. (1971) Concentrations and isotopic abundances of the rare gases in lunar matter. <i>Proc.</i> 2 nd <i>Lunar Sci. Conf.</i> 1607-1625.
Tm Yb Lu Hf Ta W ppb	0.53 3.6 0.52 3.12 0.44 110	(a) (a) (a) (a) (a)	Hubbard N.J. and Gast P.W. (1971) Chemical composition and origin of nonmare lunar basalts. <i>Proc.</i> 2 nd <i>Lunar Sci.</i> <i>Conf.</i> 999-1020.
Re ppb Os ppb Ir ppb Pt ppb Au ppb Th ppm	0.77	(a)	James O.B. and Wright T.L. (1972) Apollo 11 and 12 mare basalts and gabbros: Classification, compositional variations and possible petrogenetic relations. <i>Geol. Soc. Am. Bull.</i> 83 , 2357-2382.
U ppm technique	0.22 (a) ICP-M	(a) S	

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Table 1b. Chemical composition of 12020.

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THE CHIPPING OF LUNAR ROCK 12020

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Alexander E.C., Davis P.K. and Reynolds J.H. (1972) Raregas analysis on neutron irradiated Apollo 12 samples. *Proc.* 3rd Lunar Sci. Conf. 1787-1795.

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