63320 – 351 grams **63340** – 181 grams Shadow Soils

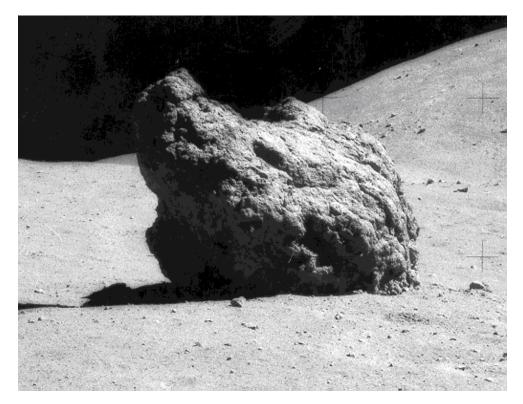


Figure 1: Shadow Rock at Apollo 16. AS16-106-17393. Rock is 5 meters across and 3 meters high (a giant sundial, ofsorts).

Introduction

Apollo 16 soil samples 63320 and 63340 were collected from a "recess" far under the shadow cast by Shadow Rock (figures 1 and 2). However, Ulrich (1981) calculates that sunlight could have reached this area during the movement of the Sun across the sky. 63320 was collected from the surface and 63340 from underneath 63320 (Sutton 1981). A "companion soil" 63501 was collected from the unshaded regolith, 15 m southeast of Shadow Rock.

Shadow Rock is located at the boundary of the Cayley Plains with the slope of Smoky mountain (figure 3). Shadow Rock is about 550 m south of the rim of North Ray Crater and the soils there are estimated to be about 50% young soil derived from North Ray Crater and 50% old soil (they have cosmic ray exposure ages of \sim 270 m.y.)

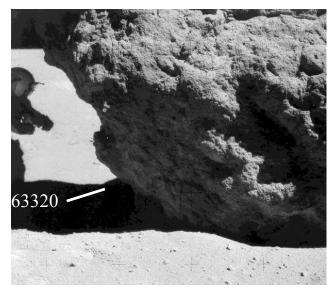


Figure 2: Shaded soil sample 63320 collected from "gopher hole" in shade. 63340 was collected from beneath 63320. AS16-160-17413.

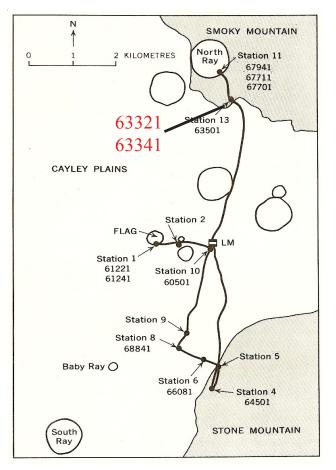


Figure 3: Map of Apollo 16 traverses showing location of station 13.

Eberhardt et al. (1976) found that ⁴⁰Ar and various volatile elements (In, Tl, Zn, Cd, Br and I) are enriched by about the same amount (~30-40%) in 63321 and 63341 compared with 63501. It is suggested that the shadowed area acted as a cold trap for volatiles in the lunar atmosphere.

Petrography

Horz et al. (1972) write "a hole more than 1 m deep and approximately 50 cm wide was observed at the south end of Shadow Rock at station 13. Because of the shadowed condition, no precise surface photography is available, but the comments of the crew about the geometry of the "gopher hole" indicate that the soil materials were permanently shielded from the Sun. After the samples were received in the LRL, samples 63320 and 63340 were placed in specially sealed containers."

63320 and 63340 are submature soils with I_s/FeO values of 47 and 54 respectively (Morris 1978). Companion soil 63501 has $I_s/FeO = 46$. Average grain

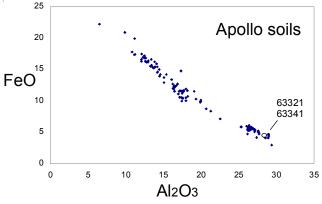


Figure 4: Samples 63321 and 63341 are among the most Al-rich lunar soils.

size is 88 microns for 63321, 80 microns for 63341 and 70 microns for 63501. Heiken et al. (1973) and Houck (1982) determined the mineralogic mode of 63321 and 63341 and compared them with nearby reference soil 63501 (they look similar). Graf (1991) reported the grain size distribution and determined the average grain size to be larger, if you take into account the > 1mm particles (figure 8 a,b).

Marvin (1972) cataloged the 4 - 10 mm coarse fine particles from these soils. Breccia samples 60017, 63335 and 63355 were chipped from Shadow Rock (Butler 1972).

Chemistry

The chemical compositions of the shadowed soils are given in tables 1 and 2 and figure 4 and 6. Eberhardt et al. (1976) compared the volatile element concentrations of 63321, 63341 and 63501 (figure 7).

Evensen et al. (1974) determined the composition of alkalis as a function of grain size. Jovanovic and Reed

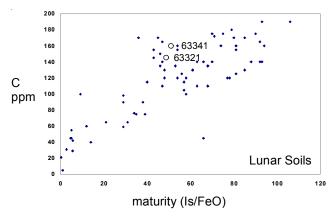


Figure 5: Samples 63321 and 63341 have relatively high carbon content (saturated?).

(1973) determined halogens, Hg, Ru, Os and U for 63321, 63341 and 63501 finding nothing unusual about their concentrations. Wanke et al. (1975) also determined halogens for 63320, finding them low in concentration. Muller (1973) and Kothari and Goel (1973) found nitrogen (~60 ppm) to be similar in all three samples. DesMarias et al. (1973) determined the carbon content (66 ppm) and Gibson and Moore (1973) found about 500 ppm sulfur (no more than elsewhere). The carbon content is relatively high (saturated?) for a submature soil (figure 5).

Cosmogenic isotopes and exposure ages

Eberhardt et al. (1976) determined the average cosmic ray exposure ages of 63321, 63241 and 63501 to be 260 m.y., 290 m.y. and 260 m.y. respectively using ⁷⁸Kr, ¹²⁴Xe and ¹²⁶ Xe measurements. Fireman et al. (1973) determined the tritium content (³H) of the shadowed soil 63321 (233 dpm/kg). Eldridge et al. (1976) determined the cosmic-ray-induced activity of $^{22}Na = 25$ and 28 dpm/kg., $^{26}Al = 69$ and 91 dpm/kg. and $^{54}Mn = 40$ and 50 dpm/kg., for 63320 and 63340, respectively.

Imarura et al. (1974) determined the cosmic–rayinduced activity of ${}^{53}Mn = 359 \text{ dpm/kg}$. The high ${}^{53}Mn$ activity could be due to lateral transport of freshly irradiated material under the boulder, or Shadow Rock may have been emplaced in its present location a short time ago (the half-life of ${}^{53}Mn$ is 3.7 m.y.).

Other Studies

Kirsten et al. (1973) and Bogard and Nyquist (1973) reported the rare gas composition of shaded soils 63321 and 63341, but didn't find anything unusual with respect to other soils. However, Eberhardt et al. (1976) discovered an enrichment of ⁴⁰Ar of 38% for 63321 and 28% for 63341 relative to 63501 (figure 6). They

Modal content o From Houck 1982	· · · · ·	41 and ref. 635	01 (90-150 micron).
	63321	63341	63501 (ref.)
Agglutinates	31.4 %	32	40.9
Basalt	-	-	-
Breccia	46.9	43.9	39.9

Breccia	46.9	43.9	39.9	
Anorthosite	0.3	0.7	-	
Norite	-	-	-	
Gabbro	-	-	-	
Plagioclase	12.2	15.7	15.1	
Pyroxene	3.6	2	0.3	
Olivine	-	0.3	0.7	
Ilmenite	-			
Glass other	5.3	5.6	2.6	

Modal content of soils 63321, 63341Imamura et al. (1974) studied 54Mn and ref. 63501 (90-150 micron).

From Heiken et al. 1973.

	63321	63341	63501 (ref.)
Agglutinates	32.6 %	40	44.6
Basalt	-	1.7	0.3
Breccia	42.5	35.5	36.3
Anorthosite	11.2	5.9	3
Norite	1.6	0.3	1.3
Gabbro	-	-	-
Plagioclase	9.6	12.6	10.3
Pyroxene	2.6	1.7	2
Olivine	-	-	-
Ilmenite	-		
Glass other	4.6	2	2.2

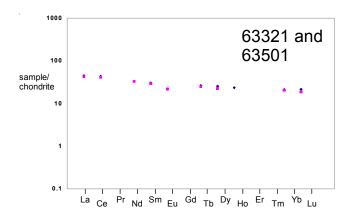


Figure 6: Normalized rare-earth-element diagram for 63321 and reference soil 63501 (data for both from Wanke et al. 1975).

found that the geometry of the shadowed area is such that reaccelerated lunar atmosphere ions, such as ⁴⁰Ar, can reach the soil, whereas the solar wind is effectively shielded.

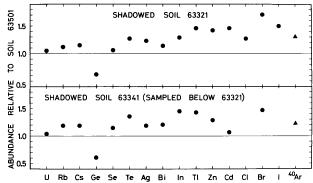


Figure 7: Volatile elements are enrich by about the same factor as 40Ar in the shadowed soils 63320 and 63240 when compared with soil 63501 (from Eberhardt et al. 1976).

Holmes et al. (1973) and Gammage and Holmes (1975) studied the influence of water vapor and liquid water on various adsorption isotherms for 63341 (figure 8). Cadenhead et al. (1977) found the surface area of 63221 and 63241 to be as expected for samples of that maturity.

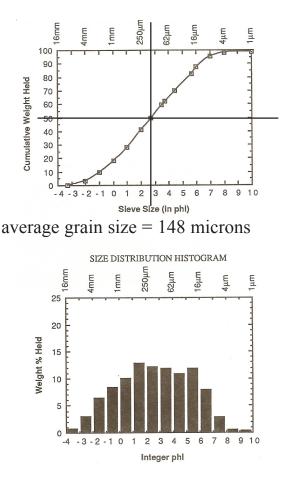
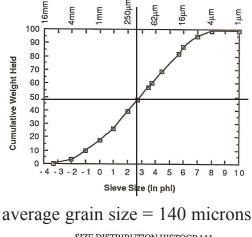


Figure 8a: Grain size analysis of 63320 (Grafe 1991, from data by McKay et al.).



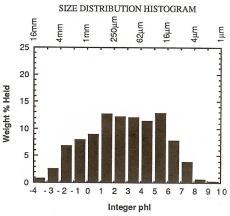


Figure 8b: Grain size distribution for 63340 (Graf 1991, data from McKay).

Table 3	U ppm	Th ppm	K ppm	technique
Eldridge et al. 1975				
63321	0.39	1.35	800	radiation
63341	0.4	1.33	790	counting
63501	0.41	1.53	728	-

Processing

Sad to say, these shadowed soils were not returned in a sealed container. They were only protected from humid spacecraft cabin air and Pacific atmosphere by the Teflon bags that they were collected in, surrounded by layers of porous beta cloth (Teflon-coated, woven, fiberglass). They would have experienced several cycles of depressurization – repressurization in the LM, one in the CM. They were sealed in Pacific air during transfer to the LRL. Finally they were exposed to the residual moisture of the "dry" N₂ glove boxes during preliminary examination.

However, only small portions of 63320 and 63340 were sieved and large portions have apparently been kept in the original Teflon collection bags (DB). In 1979, portions were split off to go into remote storage, again temporarily exposing the sample to "dry" N_2 cabinet air in the new curation facility (N_2 cabinet air has variously 10-50 ppm H₂O).

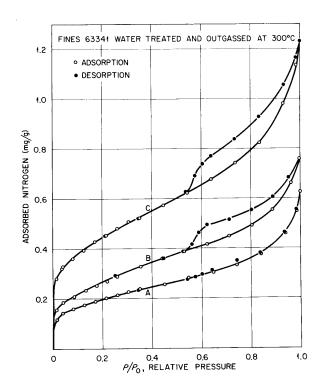
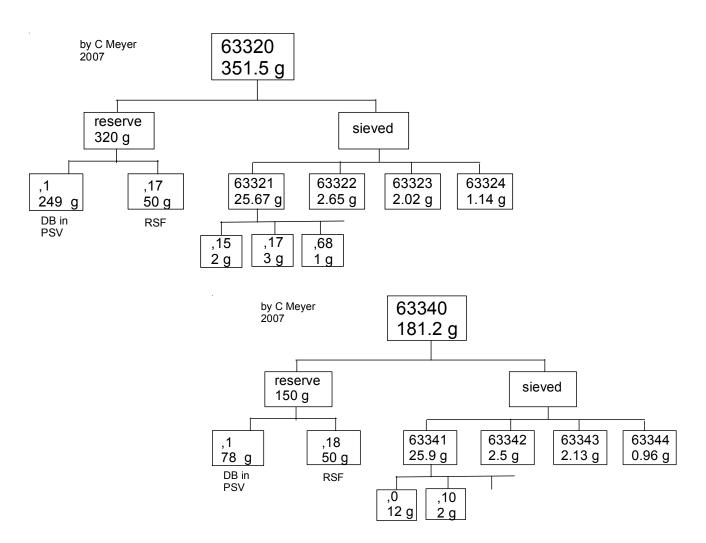


Figure 9: Adsorption isotherms of N2 for fines from 63341 (Gammage and Holmes 1975).



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desMarias D.J., Hayes J.M. and Meinshein W.G. (1974) The distribution in lunar soil of hydrogen released by pyrolysis. *Proc.* 5th *Lunar Sci. Conf.* 1811-1822.

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Evensen N.M., Murthy V.R. and Coscio M.R. (1973) Rb-Sr ages of some mare basalts and the isotopic and trace element

SiC2 %	reference weight	Brunfeldt 63321	73	Philpotts73	Krahenbuhl73		Boynton 63320	176		Evenser	173	Muller7	5	Nyquist76	Wanke	75
V 46 (a) 15 21 (a) Value	TiO2 Al2O3 FeO MnO MgO CaO Na2O K2O P2O5 S %	28.9 4.7 0.068 6.93 15.67 0.57	(a) (a) (a) (a) (a) (a)	0.096 (c)			28.1 4.76 0.065 5.8 15.4	25.2 4.63 0.067 5.97 16.5	(a) (a) (a) (a) (a)	0.11	(c)	4.8 5.7 16.23 0.58	(d) (d) (d) (d)		0.53 27.6 4.73 0.063 5.29 15.7 0.53 0.1 0.09	
Ni 311 (a) 244 250 (a) 330 (a) Zn 23 (a) 17 (b) 16.4 17 (b) 5 5.6 (b) Ga 5.5 (a) 17 (b) 16.4 17 (b) 5 5.6 (b) As	V	46	(a)				15	21	(a)						8.53	(a)
Ga 5.5 (a) 5 5.6 (b) Ga ppb 404 (b) 510 540 (b) Se 199 (b) 2.29 (c) 2.2 (e) 1.72 (c) Sr 170 (a) 179 (c) 181 (c) 172 (e) 191 (c) 198 (a) Y 170 (a) 179 (c) 181 (c) 172 (e) 191 (c) 198 (a) Y 170 (a) 179 (c) 181 (c) 172 (e) 191 (c) 198 (a) Y 170 (a) 179 (c) 181 (c) 172 (e) 191 (c) 198 (a) Y 170 (a) 179 (c) 181 (c) 172 (e) 191 (c) 198 (a) Y 16 (b) 16 (b) 172 (a) (c) 173 (a) (a	Ni	311	(a)						• •							
Se 199 (b) 2.29 (c) 2.2 (e) 1.72 (e) 1.72 (c) 1.72 (e) 198 (a) Sr 170 (a) 179 (c) 1.81 (c) 172 (e) 191 (c) 198 (a) Y 7 181 (c) 172 (e) 191 (c) 198 (a) Y 7 181 (c) 172 (e) 191 (c) 198 (a) Nb 7 16 (b) 7 (b) 7 7 (b) 7 7 (b) 7 <td>Ga Ge ppb</td> <td></td> <td>• •</td> <td></td> <td></td> <td></td> <td>5</td> <td>5.6</td> <td>(b)</td> <td></td> <td></td> <td></td> <td></td> <td></td> <td></td> <td></td>	Ga Ge ppb		• •				5	5.6	(b)							
$ \begin{array}{cccccccccccccccccccccccccccccccccccc$	Se Rb Sr Y Zr				1.8 (b										198	(a)
$ \begin{array}{cccccccccccccccccccccccccccccccccccc$	Mo Ru Rh							16	(b)							
Sn ppb 1.23 (b) Te ppb 17.3 (b) Cs ppm 0.2 (a) 0.083 (b) 0.1 (e) 0.1 (a) Ba 147 (a) 120 130 (a) 116 (c) 113 (e) 115 (a) La 10.8 (a) 10.6 9.3 (a) 9 (e) 10.5 (a) Ce 24.8 (a) 26 24 (a) 9 (e) 10.5 (a) Pr Nd	Ag ppb Cd ppb	40	(2)		· · ·))										
Cs ppm 0.2 (a) 0.083 (b) 0.1 (e) 0.1 (a) Ba 147 (a) 120 130 (a) 116 (c) 113 (e) 115 (a) La 10.8 (a) 10.6 9.3 (a) 9 (e) 10.5 (a) Ce 24.8 (a) 26 24 (a) 9 (e) 10.5 (a) Pr	Sn ppb Sb ppb	40	(a)			D)	9.0	10.5	(0)							
$\begin{array}{cccccccccccccccccccccccccccccccccccc$	Cs ppm Ba La Ce Pr	147 10.8	(a) (a)			o)	10.6	9.3	(a)	116	(c)	113	(e)		115 10.5	(a) (a)
$ \begin{array}{cccccccccccccccccccccccccccccccccccc$	Sm Eu															
Yb 3.98 (a) 3.3 3.1 (a) 3.4 (a) Lu 0.56 (a) 0.44 0.45 (a) 0.52 (a) Hf 4.8 (a) 2.8 3.3 (a) 3.95 (a) Ta 0.54 (a) 0.4 0.5 (a) 0.44 (a) W ppb 0.4 0.5 (a) 0.44 (a) Re ppb 0.669 (b) 0.4 0.5 (a) 0.44 (a) Os ppb 8.29 (b) 6.5 5 (b) 10 10 Pt ppb 8.6 (b) 4.4 3.8 (b) Silver73 10 Au ppb 8.6 (b) 4.4 3.8 (b) 1.726 (c) 1.35 (a) U ppm 0.5 (a) 0.401 (b) 0.55 0.44 (a) 0.59 (e) 0.477 (c)	Tb Dy Ho Er	4.88	(a)												6.12	(a)
Re ppb 0.669 (b) Os ppb Ir ppb 8.29 (b) 6.5 5 (b) 10 Ir ppb 8.29 (b) 6.5 5 (b) 10 Pt ppb 8.6 (b) 4.4 3.8 (b) Silver73 Th ppm 1.2 (a) 1.6 1.3 (a) 1.726 (c) 1.35 (a) U ppm 0.5 (a) 0.401 (b) 0.55 0.44 (a) 0.59 (e) 0.477 (c)	Yb Lu Hf Ta	0.56 4.8	(a) (a)				0.44 2.8	0.45 3.3	(a) (a)						0.52 3.95	(a) (a)
Pt ppb Silver73 Au ppb 8.6 (b) 4.4 3.8 (b) Silver73 Th ppm 1.2 (a) 1.6 1.3 (a) 1.726 (c) 1.35 (a) U ppm 0.5 (a) 0.401 (b) 0.55 0.44 (a) 0.59 (e) 0.477 (c)	Re ppb Os ppb						6.5	5	(þ)						10	
	Pt ppb Au ppb Th ppm U ppm	0.5	(a)	RNAA. (c)	8.6 (b 0.401 (b)))	4.4 1.6 0.55	3.8 1.3 0.44	(b) (a)			0.59	(e)	1.726 (c)		(a)

Table 1. Chemical composition of 63321.

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Table 1b. Chemical composition of 63321.

Table 1	b. Cl	nemio	cal compo	osition of 63321.	systematics in lunar lines. <i>Proc.</i> 4 th <i>Lunar Sci. Conf.</i> 1707- 1724.
reference weight SiO2 % TiO2 Al2O3	Korote C	F	Eldridge	975	Evensen N.M., Murthy V.R. and Coscio M.R. (1974) Provenance of KREEP and the exotic component: Elemental and isotopic studies of grain size fractions in lunar soils. <i>Proc.</i> 5 th <i>Lunar Sci. Conf.</i> 1401-1418.
FeO MnO MgO CaO Na2O K2O	4.84 0.56	4.72 0.53	(a) (a) 0.096	(b)	Fireman E.L., D'Amico J. and DeFelice J. (1973) Radioactivities vs. depth in Apollo 16 and 17 soil. <i>Proc.</i> 4 th <i>Lunar Sci. Conf.</i> 2131-2144.
P2O5 S % <i>sum</i> Sc ppm	8.4	7.9	(a)		Gammage and Holmes H.F. (1975) Blocking of the water- lunar fines reaction by air and water concentration effects. <i>Proc.</i> 6 th <i>Lunar Sci. Conf.</i> 3305-3316.
V Cr Co Ni	685 20 270	780 17.6 295	(a) (a) (a)		Graf J.C. (1993) Lunar Soils Grain Size Catalog. NASA Pub. 1265
Cu Zn Ga Ge ppb As					Heiken G.H., McKay D.S. and Fruland R.M. (1973) Apollo 16 soils – grain size analysis and petrography. <i>Proc.</i> 4 th Lunar Sci. Conf. 251-266.
Se Rb Sr Y Zr					Holmes H.F., Fuller E.L. and Gammage R.B. (1973) Interaction of gases with lunar materials – Apollo 12, 14 and 16 samples. <i>Proc.</i> 4 th <i>Lunar Sci. Conf.</i> 2413-2424.
Nb Mo Ru Rh Pd ppb Ag ppb					Horz F., Carrier W.D., Young J.W., Duke C.M., Nagle J.S. and Fryxell R. (1972) Apollo 16 special soils. In Apollo 16 Preliminary Science Report NASA SP315 section 7 - 25-54.
Cd ppb In ppb Sn ppb Sb ppb Te ppb					Houck K.J. (1982) Petrologic variations in Apollo 16 surface soils. <i>Proc. 13th Lunar Planet. Sci. Conf.</i> J. Geophys. Res. 87, A197-A209.
Cs ppm Ba La Ce Pr	10.2 26.5	10 27.5	(a) (a)		Imamura M., Nishiizumi K., Honda M., Finkle R.C., Arnold J.R. and Kohl C.P. (1974) Depth profiles of ⁵³ Mn in Lunar Rocks and Soils. <i>Proc.</i> 5 th Lunar Sci. Conf. 2093-2104.
Nd Sm Eu Gd Tb	4.6 1.25 1.05	4.7 1.22 0.99	(a) (a) (a)		Jovanovic S. and Reed G.W. (1973) Volatile trace elements and the characterization of the Cayley formation and the primitive lunar crust. <i>Proc.</i> 4 th <i>Lunar Sci. Conf.</i> 1313-1324.
Dy Ho Er Tm	1.00		(u)		Kerridge J.F., Kaplan I.R., Petrowski C. and Chang S. (1975) Light element geochemistry of Apollo 16 rocks and soils. <i>Geochim. Cosmochim. Acta</i> 39 , 137-162.
Yb Lu Hf Ta W ppb Re ppb	3.5 0.48 3.3 0.4	3.25 0.46 3.2 0.5	(a) (a) (a) (a)		Kirsten T., Horn P. and Kiko J. (1973) 39Ar/40Ar dating and rare gas analysis of Apollo 16 rocks and soils. <i>Proc.</i> 4 th <i>Lunar Sci. Conf.</i> 1757-1784.
Os ppb Ir ppb Pt ppb Au ppb					Kothari B.K. and Goel P.S. (1973) Nitrogen in Lunar Samples. <i>Proc.</i> 4 th Lunar Sci. Conf. 1587-1596.
Th ppm U ppm	1.6 <i>(a) IN</i> .	1.8 AA, (b)	(a) 1.35 0.39 radiation cou	(b) (b) int.	Krahenbuhl U., Ganapathy R., Morgan J.W. and Anders E. (1973a) Volatile elements in Apollo 16 samples – possible
har Sampla	Comp	andiun	2		

systematics in lunar fines. Proc. 4th Lunar Sci. Conf. 1707-

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reference weight	Brunfeld 63341	t73	Philpott	s73	Korotev81 unpublished	Krahenbul	nl73	Evense	n73	Silver73	3 Eldr	idge	75
SiO2 % TiO2 Al2O3 FeO MnO MgO CaO Na2O K2O P2O5 S % sum	0.6 29 4.53 0.067 7.3 12.45 0.57 0.147	(a) (a) (a) (a) (a) (a) (a)	0.107	(c)				0.097	(c)		0.09	5	(d)
Sc ppm V Cr Co Ni Cu Zn	7.95 34 650 19.5 345 8.3 17	(a) (a) (a) (a) (a) (a) (a)				15.5	(b)						
Ga Ge ppb	5.2	(a)				400	(b)						
As Se						216							
Rb Sr Y	1.7 140		2.49 181	(c) (c)		1.9	(b) (b)	2.05 180	(c) (c)				
Zr Nb Mo Ru Rh Pd ppb													
Ag ppb Cd ppb						7.6 57.5	(b) (b)						
In ppb	45	(a)				01.0	(0)						
Sn ppb Sb ppb						1.23	(b)						
Te ppb Cs ppm	0.05	(a)				18.5 0.086	(b) (b)						
Ва	92	(a)				0.000	(0)	97.5	(C)				
La Ce	11.2 25.2	(a) (a)											
Pr		()											
Nd Sm	4.45	(a)											
Eu	1.42	(a)											
Gd Tb	0.93	(a)											
Dy Ho	6.18 1.5	(a)											
Er	1.5	(a)											
Tm Yb	3.87	(a)											
Lu	0.56	(a)											
Hf Ta	3.5 0.45	(a)											
W ppb	0.45	(a)											
Re ppb						0.741	(b)						
Os ppb Ir ppb						11.1	(b)						
Pt ppb Au ppb						7.08	(b)						
Th ppm	1	(a)								1.8	(c) 1.33		(d)
U ppm technique.	0.4 : (a) INAA	(a) A. <i>(b</i>)) RNAA	(c) I	DMS, (d) radia	0.398 ntion cout.	(b)			0.617	(c) 0.4		(d)
		, (~)	,		- / (-) /								

Table 2. Chemical composition of 63340.

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Krahenbuhl U., Ganapathy R., Morgan J.W. and Anders E. (1973b) Volatile elements in Apollo 16 samples – Implications for highland volcanism and accretion history of the Moon. *Proc.* 4th *Lunar Sci. Conf.* 1325-1348.

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